

### Correction

Thermodynamic Equilibrium of Diluted SO<sub>2</sub> Absorption into Na<sub>2</sub>SO<sub>4</sub> or H<sub>2</sub>SO<sub>4</sub> Electrolyte Solutions. T. Hunger, F. Lapique, and A. Storck, *J. Chem. Eng. Data* 1990, 35, 453–463.

In Table IV, the volumes  $V_b$  and  $V_c$  should be expressed in dm<sup>3</sup> mol<sup>-1</sup> and not in dm<sup>3</sup> kg<sup>-1</sup>. In relation A3, the expression  $\Gamma_{ref}$  and not  $\Gamma_{rel}$  should be inserted. In relation A5, an exponent was omitted in the expression for the Debye–Hückel coefficient, and the correct expression should be

$$A = \frac{1}{2.303} \left[ \frac{e}{(DkT)^{1/2}} \right]^3 \left( \frac{2\pi d_0 N_A}{1000} \right)^{1/2}$$

instead of

$$A = \frac{1}{2.303} \left[ \frac{e}{(DkT)^{1/2}} \right] \left( \frac{2\pi d_0 N_A}{1000} \right)^{1/2}$$

An error in the right-hand term of relation A10 was made, and this relation has to be replaced by

$$T_{\text{ion,ion}(j+4)} = -z^2(j+4)B_{\text{sum}}/4I^2 + 2\sum_{l=1}^2 m(l+2)B(l,j)$$

The dimensionless dielectric constant should be defined as (A12)

$$\text{DIV} = \left[ 1 + \frac{\alpha(2)m(2)}{V_m} \right]^{-1}$$

A parenthesis is missing in the numerator of the last term of relation A15b, related to the ion–molecule interaction term for SO<sub>2</sub>; the correct expression is

$$T_{\text{ion-mol}(2)} = \sum L \text{DIV} \left[ \frac{-1.5V_b(2) V_{cT}}{V_{fc}^2} + \frac{V_f + 0.5V_{cT} \left( \frac{V_b(2)}{V_m} - \frac{\text{DIV}(V_b(2) + \alpha(2))}{V_m} \right)}{V_{fc}} \right]$$

Finally, the last term for the ion–molecule interaction contribution of H<sub>2</sub>O should be proportional to  $\sum LM_w$ ; therefore, the actual relation A15c is

$$T_{\text{ion-ion}(1)} = \frac{2AM_w}{3} \frac{I^{1.5}}{1 + 1.2I^{1/2}} - 2M_w \sum_{l=1}^2 m(l+2) \sum_{j=1}^4 m(j+4) B_w(l,j) - \sum LM_w \left[ \frac{-\alpha(2)m(2)\text{DIV}^2}{d_0 V_m^2} \times \frac{V_f + 0.5V_{cT}}{V_{fc}} + \frac{1.5V_{cT}\text{DIV}}{d_0 V_{fc}^2} - \frac{1.5V_{cT}}{d_0 V_{fc}^2} \right]$$